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Answer Key Chapter 18 -

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OpenStax balanced: $2\text{Ag}(s) + \text{Cu}^{2+}$

$+ (aq) \rightarrow 2\text{Ag} + (aq) + \text{Cu}(s)$ E°

$= E^\circ_{\text{ox}}(\text{Ag}) + E^\circ_{\text{red}}(\text{Cu}^{2+}) =$

$- 0.80\text{V} + 0.34\text{V} = - 0.46\text{V}$. Not

spontaneous. ox: $[\text{Ni}(s) \rightarrow \text{Ni}^{2+} + +$

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$2e^-] \times 5$; red: $[5e^- + 8H^+ + MnO_4^- \rightarrow Mn^{2+} + 4H_2O] \times 2$.
balanced: $5Ni(s) +$

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balanced: $2Ag(s) + Cu^{2+}(aq) \rightarrow$
 $2Ag^+(aq) + Cu(s)$ $E^\circ = E^\circ$

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ox(Ag) + E ° red(Cu²⁺) = -
0.80V + 0.34V = - 0.46V. Not
spontaneous. ox: [Ni(s) → Ni²⁺ +
2e⁻] × 5; red: [5e⁻ + 8H⁺ +
MnO₄⁻ → Mn²⁺ + 4H₂O] × 2.
balanced: 5Ni(s) + 16H⁺(aq) +
2MnO₄⁻(aq) → 5Ni²⁺(aq) +
2Mn²⁺(aq) + 8H₂O.

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Because protons and neutrons reside in the nucleus of atoms, they are called nucleons. 16.

There are two forces among the particles within the nucleus. The first, called the electrostatic force, is the force between electrically charged particles. The second

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force,

~~Chapter 18 Nuclear Chemistry~~
a) the energy given off when reactants collide b) *an energy barrier between reactants and products the heat released in a reaction* c) generally very high

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for a reaction that takes place
rapidly

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Chapter 18 Amines and Amides.
18.1 Multiple Choice Questions. 1)
The compound $\text{CH}_3 - \text{CH}_2 - \text{NH} - \text{CH}_3$ is classified as a. A) primary amine. B) secondary amine. C)

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tertiary amine. D) quaternary amine. E) hydrated amine.

Answer: B. Objective: 18.1. Global Outcomes: GO2. 2) The compound $\text{CH}_3 - \text{CH}_2 - \text{NH}_2$ is classified as a. A) primary amine. B) secondary amine.

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