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Science 1-5.  
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Salts Quiz **ACIDS**  
**BASES \u0026**  
**SALTS-FULL**  
**CHAPTER ||**  
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salts chapter 5

**QUESTION**

**ANSWERS full**

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**QUESTIONS \u0026**

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*Acids and Bases*

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*Questions pH*

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Hindi **Questions**  
**and answers on**  
**acid, base, and**  
**salt** ||

Answers,  
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**Wonders Acids**

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~~Answers~~

Solution: The reaction of the acid + base gives a product of salt + water, which is considered as neutralization reaction.

Examples:  $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$

~~NCERT Solutions  
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~~Chapter 2 Acids,  
bases and salt~~  
Class 10 Science  
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Class 10 Science  
Chapter 2 Acids  
Bases and Salts  
with Answers  
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Bases and Salts

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## Acids Bases

Extra Questions  
Very Short Answer  
Type. Question 1.  
Two solutions A  
and B have pH  
values of 5 and 8  
respectively. Which  
solution will be  
basic in nature ...

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Salts Class 10  
Extra Questions  
with ...~~



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Ans:- Salts are classified into four categories on the basis of the strength of acid and base. they are:

- Salt of strong acid and strong base.e.g NaCl, KCl;
- Salt of strong acid and weak base. e.g.  $\text{NH}_4\text{Cl}$ ,  $\text{NH}_4\text{I}$ ;
- Salt of weak acid and strong base.

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## Acids Bases

e.g.  $\text{CH}_3\text{COONa}$ ,  
 $\text{COO}^- \text{Na}^+$  + Salt of  
weak acid and  
weak base. e.g.  $\text{CH}_3\text{COO}^- \text{NH}_4^+$  +  
How is common  
salt found in  
nature?

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## Acids Bases

Acids, Bases &  
Salts. 8.1 Acids,  
Bases & Oxides;  
8.2 Salts &  
Chemical Analysis;  
9. The Periodic  
Table ...

~~Acids, Bases &  
Oxides | CIE IGCSE  
Chemistry | MCQ &  
Answers~~

2. Bases react with  
. acids to produce

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## Acids Bases

salts and water;  
water to produce  
acids and salts;  
salts to produce  
acids and water;  
neither acids, salts,  
nor water

~~Acids Bases and  
Salts Multiple  
Choice Questions  
and Answers~~

The salts of strong  
acids and weak

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And Salts  
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bases give acidic solution having pH less than 7.

Example,  $\text{NH}_4\text{Cl}$ ,  
Ammonium

Chloride will have pH less than 7. The salts of weak acids and strong bases give basic solution having pH more than 7. Example,  $\text{Na}_2\text{CO}_3$ , Sodium Carbonate will

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have pH more than  
7. Question 7.

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~~Important~~

~~Questions with ...~~

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## Education

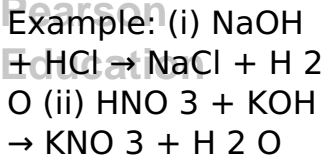
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Answer Key  
Teacher  
Worksheets~~

Answer A reaction  
in which an acid  
and base react  
with each other to  
give a salt and

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## Acids Bases

water is termed as neutralization reaction. For



15. Give two important uses of washing soda and baking soda.

Answer Two important uses of washing soda are:

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for Class 10th  
Science: Ch 2  
Acids, Bases ...

Answer. (a) HCl is acid and NaOH is base whose combination forms the common salt. Its formula is NaCl (Sodium chloride). It is obtained from sea water. (b) Rock

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salt is the common name for the mineral "halite". Its chemical formula is  $\text{NaCl}$ . It may be white or light blue or yellow depending upon impurities present in it.

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phenylamine  
aniline as an amine  
chemguide.

~~Acids Bases And  
Salts Note Taking  
Answers~~

(A) inorganic acid

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## Acids Bases

(B) organic acid (C)  
soft organic acid  
(D) strong  
inorganic acid.

Answer. Answer:

(a) 5. When milk of  
magnesia reacts  
with acetic acid it  
produces (A) basic  
salts (B) acidic  
salts (C) neutral  
salt (D) complex  
salt. Answer.

Answer: (c) 6.

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Which of the following phenomena will occur when a small amount of acid is added to water ...

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Turmeric is a natural indicator.

On adding its paste to acid and base separately, which colours would be observed. (a)

Yellow in both acid and base (b) Yellow in acid and red in base (c) Pink in

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acid and yellow in  
base (d) Red in  
acid and blue in  
base. Answer.

Answer: (b) Yellow  
in acid and red in  
base

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Chapter 5 Acids,  
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MCQ Questions for  
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with Answers were prepared based on the latest exam pattern. We have provided Acids, Bases and Salts Class 7 Science MCQs Questions with Answers to help students understand the concept very well. Acids, Bases and Salts Class 7 MCQs

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Questions with  
Answers. Choose  
the correct answer:  
Question 1.

## Education

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Acids, Bases and  
Salts In this first  
quiz on GCSE  
Chemistry we take  
a look at one vital

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## Acids Bases

part of the subject - acids, alkalis, bases and salts. In it we find out the difference between alkalis and bases, how acids and alkalis react with one another, how neutralisation can produce salts and the relevance of pH values.



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~~GCSE Acids, Bases  
and Salts | Revise  
pH Values and  
Alkalis~~

Acids, bases and salts An acid is a compound that is defined by its physical and chemical properties. Acids taste sour and react with metals and polyatomic

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ions called  
carbonates. A  
carbonate is a  
charged cluster of  
Carbon and  
Oxygen atoms.

~~Acids, bases and  
salts. 7th Grade  
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help students  
understand the  
concept very well.

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Chapter 2 Acids,  
Bases ...~~

Acid + Base --->  
Salt + Water.

Question 3: Rain  
becomes acidic  
due to. carbon

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dioxide, sulphur dioxide, nitrogen dioxide, all of the above. Correct

Option is : 4.

Solution : Rain becomes acidic because of carbon dioxide, sulphur dioxide and nitrogen dioxide. They dissolve in rain drops to form carbonic acid,

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Answers  
sulphuric acid and  
nitric acid ...

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